

## **Application Note**

Date: 1992.6.16 No. 720047H-E

## Analysis of vitamin B1 in health drink

Vitamin B1 in nutritional drink was analyzed by using the post column derivatization method with a fluorescence detector. Potassium ferricyanide was used as a reagent. The chromatograms obtained by injecting 10ul samples of drink (1) and (2) were shown in Fig.1 and Fig.2.

## **Conditions:**

Detector: Fluorescence detector Wavelength: Ex. 372nm, Em. 460nm

Sensitivity: x 100, x 8

Column: Finepak SIL C18S

Eluent: 1mM 1-Hexanesulfonic acid

sodium salt + 0.1M KH<sub>2</sub>PO<sub>4</sub> (pH 3.6) / CH<sub>2</sub>OH (90/10)

Flow rate: 0.5 ml/min

Temperature : 70 degree celsius Reagent :  $10\% \text{ NaOH} + 0.01\% \text{ K}_3[\text{Fe}(\text{CN})_6]$ 

Reagent flow rate: 0.5 ml/min

Reaction coil: 0.5mm diameter x 5m long

Tonic drink (2)

Reaction temperature: 70 degree celsius Sample: Tonic drink (1)

